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12-(2-Methoxyphenyl)-9,9-dimethyl-8,9-dihydro-12H-benzo[a]xanthen-11(10H)-one

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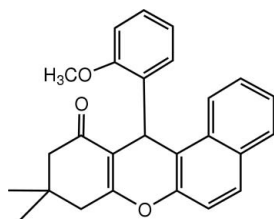
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Key indicators: single-crystal X-ray study; $T = 113$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.049; wR factor = 0.122; data-to-parameter ratio = 14.5.

The title compound, $\text{C}_{26}\text{H}_{24}\text{O}_3$, was synthesized *via* the coupling of 2-methoxybenzaldehyde, 2-naphthol and 5,5-dimethylcyclohexane-1,3-dione. The pyran ring adopts a boat conformation, while the cyclohexenone ring is in an envelope conformation. The 2-methoxyphenyl ring is almost perpendicular to the plane through the four C atoms of the pyran ring [dihedral angle = $88.76(9)^\circ$].

Related literature

For the antiviral activity of xanthenes and benzoxanthenes, see: Lambert *et al.* (1997).



Experimental

Crystal data

$\text{C}_{26}\text{H}_{24}\text{O}_3$	$V = 1987.4(5) \text{ \AA}^3$
$M_r = 384.45$	$Z = 4$
Monoclinic, $P2_1/n$	Mo $K\alpha$ radiation
$a = 7.8454(11) \text{ \AA}$	$\mu = 0.08 \text{ mm}^{-1}$
$b = 22.670(3) \text{ \AA}$	$T = 113 \text{ K}$
$c = 11.3100(13) \text{ \AA}$	$0.36 \times 0.28 \times 0.20 \text{ mm}$
$\beta = 98.893(4)^\circ$	

Data collection

Rigaku Saturn CCD area-detector diffractometer	12038 measured reflections
Absorption correction: multi-scan (<i>CrystalClear</i> ; Rigaku/MS, 2005)	3860 independent reflections
$T_{\min} = 0.970$, $T_{\max} = 0.985$	3274 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.035$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.049$	266 parameters
$wR(F^2) = 0.122$	H-atom parameters constrained
$S = 1.09$	$\Delta\rho_{\text{max}} = 0.29 \text{ e \AA}^{-3}$
3860 reflections	$\Delta\rho_{\text{min}} = -0.21 \text{ e \AA}^{-3}$

Data collection: *CrystalClear* (Rigaku/MS, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: NC2168).

References

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supplementary materials

Acta Cryst. (2010). E66, o119 [doi:10.1107/S1600536809052210]

12-(2-Methoxyphenyl)-9,9-dimethyl-8,9-dihydro-12*H*-benzo[*a*]xanthen-11(10*H*)-one

D.-L. Li and L.-H. Wang

Experimental

To a mixture of 2-naphthol (1.0 mmol), 2-methoxybenzaldehyde (1.0 mmol), and 5,5-dimethylcyclohexane-1,3-dione (1.1 mmol) strontium trifluoromethanesulfonate (0.1 mmol) in 1,2-dichloroethane (2 ml) was added. The mixture was stirred at 80 °C for 5 h and the progress of the reaction was monitored by thin layer chromatography. After completion of the reaction, 5 ml of water were added and the product was extracted three times with 10 ml of ethyl acetate. The organic layer was dried over MgSO₄ filtered off and the solvent was evaporated. The crude was product dissolved in ethylacetate and purified by flash chromatography on silica gel. The solvent was evaporated and the product was dissolved in ethanol. A single crystal was obtained by slow evaporation of the solvent from a solution in ethanol.

Refinement

All H atoms were included in the refinement in the riding model approximation, with C—H = 0.95–1.00 Å, and with $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{C})$ or $1.5 U_{\text{eq}}(\text{C})$ for methyl H atoms.

Figures

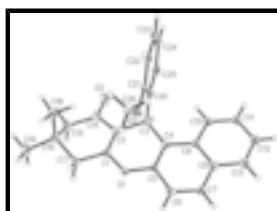


Fig. 1. A view of the molecular structure of (I), showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level.

12-(2-Methoxyphenyl)-9,9-dimethyl-9,10-dihydro-8*H*- benzo[*a*]xanthen-11(12*H*)-one

Crystal data

C₂₆H₂₄O₃

$M_r = 384.45$

Monoclinic, $P2_1/n$

$a = 7.8454$ (11) Å

$b = 22.670$ (3) Å

$c = 11.3100$ (13) Å

$\beta = 98.893$ (4)°

$V = 1987.4$ (5) Å³

$Z = 4$

$F(000) = 816$

$D_x = 1.285$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71070$ Å

Cell parameters from 3867 reflections

$\theta = 1.8$ – 27.2 °

$\mu = 0.08$ mm⁻¹

$T = 113$ K

Block, colorless

$0.36 \times 0.28 \times 0.20$ mm

Data collection

Rigaku Saturn CCD area-detector diffractometer	3860 independent reflections
Radiation source: rotating anode confocal	3274 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.035$
Detector resolution: 7.31 pixels mm^{-1}	$\theta_{\text{max}} = 26.0^\circ$, $\theta_{\text{min}} = 2.0^\circ$
ω and φ scans	$h = -9 \rightarrow 8$
Absorption correction: multi-scan (<i>CrystalClear</i> ; Rigaku/MSO, 2005)	$k = -27 \rightarrow 20$
$T_{\text{min}} = 0.970$, $T_{\text{max}} = 0.985$	$l = -13 \rightarrow 13$
12038 measured reflections	

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.049$	H-atom parameters constrained
$wR(F^2) = 0.122$	$w = 1/[\sigma^2(F_o^2) + (0.0629P)^2 + 0.2114P]$
$S = 1.09$	where $P = (F_o^2 + 2F_c^2)/3$
3860 reflections	$(\Delta/\sigma)_{\text{max}} = 0.001$
266 parameters	$\Delta\rho_{\text{max}} = 0.29 \text{ e } \text{\AA}^{-3}$
0 restraints	$\Delta\rho_{\text{min}} = -0.21 \text{ e } \text{\AA}^{-3}$
Primary atom site location: structure-invariant direct methods	Extinction correction: <i>SHELXL</i> , $F_c^* = kFc[1 + 0.001 \times Fc^2 \lambda^3 / \sin(2\theta)]^{-1/4}$
	Extinction coefficient: 0.024 (2)

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.32998 (13)	0.05276 (4)	0.89201 (9)	0.0251 (3)
O2	-0.07739 (15)	0.20650 (5)	0.81569 (13)	0.0444 (4)
O3	0.19539 (13)	0.05495 (5)	0.63209 (9)	0.0249 (3)

C1	0.2801 (2)	0.11049 (7)	0.87702 (13)	0.0239 (4)
C2	0.11615 (19)	0.12678 (7)	0.84090 (13)	0.0237 (4)
C3	-0.0248 (2)	0.08279 (6)	0.79729 (13)	0.0226 (3)
H3	-0.1274	0.0930	0.8358	0.027*
C4	0.03259 (19)	0.02110 (7)	0.83795 (12)	0.0220 (3)
C5	0.20193 (19)	0.00984 (7)	0.88233 (12)	0.0229 (3)
C6	0.2635 (2)	-0.04641 (7)	0.92099 (13)	0.0259 (4)
H6	0.3813	-0.0518	0.9543	0.031*
C7	0.1525 (2)	-0.09314 (7)	0.91012 (13)	0.0266 (4)
H7	0.1936	-0.1312	0.9357	0.032*
C8	-0.0236 (2)	-0.08537 (7)	0.86104 (13)	0.0249 (4)
C9	-0.08557 (19)	-0.02749 (7)	0.82836 (12)	0.0238 (4)
C10	-0.2640 (2)	-0.02089 (7)	0.78388 (13)	0.0272 (4)
H10	-0.3097	0.0174	0.7653	0.033*
C11	-0.3711 (2)	-0.06874 (8)	0.76733 (14)	0.0328 (4)
H11	-0.4899	-0.0632	0.7372	0.039*
C12	-0.3080 (2)	-0.12620 (8)	0.79428 (15)	0.0342 (4)
H12	-0.3825	-0.1593	0.7792	0.041*
C13	-0.1390 (2)	-0.13381 (7)	0.84225 (14)	0.0311 (4)
H13	-0.0978	-0.1724	0.8635	0.037*
C14	0.0722 (2)	0.18961 (7)	0.84212 (15)	0.0305 (4)
C15	0.2164 (2)	0.23259 (7)	0.88335 (17)	0.0352 (4)
H15A	0.2239	0.2383	0.9708	0.042*
H15B	0.1868	0.2711	0.8444	0.042*
C16	0.3943 (2)	0.21341 (7)	0.85643 (16)	0.0322 (4)
C17	0.4295 (2)	0.15108 (7)	0.90742 (14)	0.0275 (4)
H17A	0.5303	0.1344	0.8761	0.033*
H17B	0.4594	0.1537	0.9956	0.033*
C18	0.3939 (2)	0.21340 (8)	0.72089 (17)	0.0412 (5)
H18A	0.3564	0.2521	0.6880	0.062*
H18B	0.3146	0.1830	0.6836	0.062*
H18C	0.5106	0.2051	0.7044	0.062*
C19	0.5342 (2)	0.25523 (8)	0.9164 (2)	0.0473 (5)
H19A	0.6479	0.2401	0.9058	0.071*
H19B	0.5274	0.2581	1.0020	0.071*
H19C	0.5173	0.2944	0.8798	0.071*
C20	-0.07830 (19)	0.08847 (6)	0.66195 (13)	0.0217 (3)
C21	0.03606 (19)	0.07426 (6)	0.58126 (13)	0.0218 (3)
C22	-0.0157 (2)	0.08026 (7)	0.45887 (13)	0.0250 (4)
H22	0.0609	0.0697	0.4050	0.030*
C23	-0.1784 (2)	0.10161 (7)	0.41493 (14)	0.0279 (4)
H23	-0.2132	0.1054	0.3310	0.033*
C24	-0.2910 (2)	0.11751 (7)	0.49250 (14)	0.0277 (4)
H24	-0.4017	0.1330	0.4625	0.033*
C25	-0.2392 (2)	0.11041 (6)	0.61498 (14)	0.0245 (4)
H25	-0.3169	0.1210	0.6681	0.029*
C26	0.3182 (2)	0.04429 (7)	0.55369 (15)	0.0296 (4)
H26A	0.2763	0.0126	0.4978	0.044*
H26B	0.4286	0.0327	0.6007	0.044*

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H26C 0.3341 0.0803 0.5087 0.044*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0227 (6)	0.0247 (6)	0.0265 (6)	-0.0001 (4)	-0.0007 (5)	0.0011 (4)
O2	0.0281 (7)	0.0341 (7)	0.0693 (9)	0.0077 (5)	0.0024 (6)	-0.0136 (6)
O3	0.0208 (6)	0.0320 (6)	0.0223 (6)	0.0042 (5)	0.0041 (5)	-0.0003 (4)
C1	0.0275 (9)	0.0259 (8)	0.0182 (7)	0.0012 (7)	0.0032 (6)	-0.0024 (6)
C2	0.0232 (8)	0.0276 (8)	0.0203 (8)	0.0011 (6)	0.0037 (6)	-0.0042 (6)
C3	0.0216 (8)	0.0261 (8)	0.0205 (8)	0.0023 (6)	0.0046 (6)	-0.0024 (6)
C4	0.0246 (9)	0.0281 (8)	0.0137 (7)	0.0012 (6)	0.0039 (6)	-0.0012 (6)
C5	0.0234 (8)	0.0280 (8)	0.0173 (7)	-0.0028 (6)	0.0032 (6)	-0.0013 (6)
C6	0.0251 (9)	0.0316 (9)	0.0201 (8)	0.0023 (7)	0.0007 (7)	-0.0005 (6)
C7	0.0336 (9)	0.0256 (8)	0.0210 (8)	0.0021 (7)	0.0055 (7)	0.0011 (6)
C8	0.0292 (9)	0.0292 (8)	0.0171 (7)	-0.0016 (7)	0.0061 (7)	-0.0012 (6)
C9	0.0253 (9)	0.0314 (8)	0.0151 (7)	-0.0016 (7)	0.0047 (6)	-0.0018 (6)
C10	0.0266 (9)	0.0344 (9)	0.0209 (8)	-0.0023 (7)	0.0050 (7)	-0.0002 (6)
C11	0.0286 (9)	0.0461 (10)	0.0239 (8)	-0.0055 (8)	0.0049 (7)	0.0000 (7)
C12	0.0367 (10)	0.0386 (10)	0.0282 (9)	-0.0131 (8)	0.0073 (8)	-0.0026 (7)
C13	0.0382 (10)	0.0309 (9)	0.0256 (8)	-0.0041 (7)	0.0090 (7)	0.0009 (7)
C14	0.0265 (9)	0.0308 (9)	0.0346 (9)	0.0035 (7)	0.0060 (7)	-0.0067 (7)
C15	0.0310 (10)	0.0276 (9)	0.0471 (11)	0.0017 (7)	0.0061 (8)	-0.0097 (7)
C16	0.0278 (9)	0.0250 (8)	0.0438 (10)	0.0013 (7)	0.0054 (8)	-0.0029 (7)
C17	0.0245 (9)	0.0288 (9)	0.0289 (8)	-0.0014 (7)	0.0026 (7)	-0.0037 (7)
C18	0.0436 (11)	0.0340 (10)	0.0483 (11)	0.0069 (8)	0.0148 (9)	0.0105 (8)
C19	0.0320 (11)	0.0333 (10)	0.0761 (15)	-0.0040 (8)	0.0066 (10)	-0.0085 (9)
C20	0.0224 (8)	0.0193 (7)	0.0231 (8)	-0.0006 (6)	0.0024 (6)	-0.0003 (6)
C21	0.0210 (8)	0.0207 (7)	0.0228 (8)	-0.0007 (6)	0.0008 (6)	0.0004 (6)
C22	0.0275 (9)	0.0256 (8)	0.0220 (8)	-0.0028 (6)	0.0045 (7)	0.0006 (6)
C23	0.0303 (9)	0.0304 (9)	0.0212 (8)	-0.0051 (7)	-0.0016 (7)	0.0055 (6)
C24	0.0225 (9)	0.0279 (8)	0.0306 (9)	-0.0012 (7)	-0.0027 (7)	0.0071 (7)
C25	0.0224 (8)	0.0235 (8)	0.0281 (8)	0.0002 (6)	0.0053 (7)	0.0009 (6)
C26	0.0253 (9)	0.0351 (9)	0.0303 (9)	0.0033 (7)	0.0106 (7)	0.0013 (7)

Geometric parameters (\AA , $^\circ$)

O1—C1	1.3689 (18)	C14—C15	1.511 (2)
O1—C5	1.3907 (18)	C15—C16	1.537 (2)
O2—C14	1.227 (2)	C15—H15A	0.9900
O3—C21	1.3647 (18)	C15—H15B	0.9900
O3—C26	1.4274 (17)	C16—C19	1.528 (2)
C1—C2	1.340 (2)	C16—C18	1.532 (2)
C1—C17	1.488 (2)	C16—C17	1.535 (2)
C2—C14	1.466 (2)	C17—H17A	0.9900
C2—C3	1.514 (2)	C17—H17B	0.9900
C3—C4	1.519 (2)	C18—H18A	0.9800
C3—C20	1.529 (2)	C18—H18B	0.9800
C3—H3	1.0000	C18—H18C	0.9800

C4—C5	1.370 (2)	C19—H19A	0.9800
C4—C9	1.433 (2)	C19—H19B	0.9800
C5—C6	1.409 (2)	C19—H19C	0.9800
C6—C7	1.365 (2)	C20—C25	1.384 (2)
C6—H6	0.9500	C20—C21	1.412 (2)
C7—C8	1.418 (2)	C21—C22	1.388 (2)
C7—H7	0.9500	C22—C23	1.384 (2)
C8—C13	1.418 (2)	C22—H22	0.9500
C8—C9	1.428 (2)	C23—C24	1.385 (2)
C9—C10	1.420 (2)	C23—H23	0.9500
C10—C11	1.367 (2)	C24—C25	1.391 (2)
C10—H10	0.9500	C24—H24	0.9500
C11—C12	1.410 (2)	C25—H25	0.9500
C11—H11	0.9500	C26—H26A	0.9800
C12—C13	1.364 (2)	C26—H26B	0.9800
C12—H12	0.9500	C26—H26C	0.9800
C13—H13	0.9500		
C1—O1—C5	118.04 (12)	C16—C15—H15B	108.6
C21—O3—C26	117.06 (11)	H15A—C15—H15B	107.6
C2—C1—O1	122.93 (14)	C19—C16—C18	109.51 (15)
C2—C1—C17	125.79 (14)	C19—C16—C17	109.22 (14)
O1—C1—C17	111.27 (13)	C18—C16—C17	110.37 (13)
C1—C2—C14	118.75 (14)	C19—C16—C15	110.43 (14)
C1—C2—C3	122.43 (14)	C18—C16—C15	109.70 (15)
C14—C2—C3	118.82 (13)	C17—C16—C15	107.60 (13)
C2—C3—C4	109.91 (13)	C1—C17—C16	113.13 (13)
C2—C3—C20	110.14 (12)	C1—C17—H17A	109.0
C4—C3—C20	113.79 (12)	C16—C17—H17A	109.0
C2—C3—H3	107.6	C1—C17—H17B	109.0
C4—C3—H3	107.6	C16—C17—H17B	109.0
C20—C3—H3	107.6	H17A—C17—H17B	107.8
C5—C4—C9	117.64 (14)	C16—C18—H18A	109.5
C5—C4—C3	120.53 (14)	C16—C18—H18B	109.5
C9—C4—C3	121.81 (14)	H18A—C18—H18B	109.5
C4—C5—O1	123.12 (14)	C16—C18—H18C	109.5
C4—C5—C6	123.30 (14)	H18A—C18—H18C	109.5
O1—C5—C6	113.58 (13)	H18B—C18—H18C	109.5
C7—C6—C5	119.42 (15)	C16—C19—H19A	109.5
C7—C6—H6	120.3	C16—C19—H19B	109.5
C5—C6—H6	120.3	H19A—C19—H19B	109.5
C6—C7—C8	120.55 (15)	C16—C19—H19C	109.5
C6—C7—H7	119.7	H19A—C19—H19C	109.5
C8—C7—H7	119.7	H19B—C19—H19C	109.5
C7—C8—C13	121.56 (15)	C25—C20—C21	117.83 (14)
C7—C8—C9	119.20 (14)	C25—C20—C3	120.50 (13)
C13—C8—C9	119.24 (15)	C21—C20—C3	121.61 (13)
C10—C9—C8	117.79 (14)	O3—C21—C22	124.02 (13)
C10—C9—C4	122.44 (15)	O3—C21—C20	115.66 (13)
C8—C9—C4	119.75 (14)	C22—C21—C20	120.32 (14)

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C11—C10—C9	121.07 (16)	C23—C22—C21	120.27 (14)
C11—C10—H10	119.5	C23—C22—H22	119.9
C9—C10—H10	119.5	C21—C22—H22	119.9
C10—C11—C12	121.03 (16)	C22—C23—C24	120.44 (14)
C10—C11—H11	119.5	C22—C23—H23	119.8
C12—C11—H11	119.5	C24—C23—H23	119.8
C13—C12—C11	119.36 (16)	C23—C24—C25	118.95 (15)
C13—C12—H12	120.3	C23—C24—H24	120.5
C11—C12—H12	120.3	C25—C24—H24	120.5
C12—C13—C8	121.38 (16)	C20—C25—C24	122.15 (14)
C12—C13—H13	119.3	C20—C25—H25	118.9
C8—C13—H13	119.3	C24—C25—H25	118.9
O2—C14—C2	121.23 (15)	O3—C26—H26A	109.5
O2—C14—C15	121.08 (15)	O3—C26—H26B	109.5
C2—C14—C15	117.61 (14)	H26A—C26—H26B	109.5
C14—C15—C16	114.58 (13)	O3—C26—H26C	109.5
C14—C15—H15A	108.6	H26A—C26—H26C	109.5
C16—C15—H15A	108.6	H26B—C26—H26C	109.5
C14—C15—H15B	108.6		
C5—O1—C1—C2	-7.5 (2)	C10—C11—C12—C13	2.7 (2)
C5—O1—C1—C17	171.74 (11)	C11—C12—C13—C8	-2.6 (2)
O1—C1—C2—C14	172.34 (13)	C7—C8—C13—C12	-179.56 (14)
C17—C1—C2—C14	-6.8 (2)	C9—C8—C13—C12	-0.4 (2)
O1—C1—C2—C3	-8.5 (2)	C1—C2—C14—O2	-176.29 (15)
C17—C1—C2—C3	172.37 (13)	C3—C2—C14—O2	4.5 (2)
C1—C2—C3—C4	18.08 (19)	C1—C2—C14—C15	0.6 (2)
C14—C2—C3—C4	-162.79 (13)	C3—C2—C14—C15	-178.59 (14)
C1—C2—C3—C20	-108.08 (16)	O2—C14—C15—C16	-152.27 (17)
C14—C2—C3—C20	71.06 (16)	C2—C14—C15—C16	30.9 (2)
C2—C3—C4—C5	-13.38 (18)	C14—C15—C16—C19	-172.21 (15)
C20—C3—C4—C5	110.68 (15)	C14—C15—C16—C18	67.01 (19)
C2—C3—C4—C9	168.47 (12)	C14—C15—C16—C17	-53.1 (2)
C20—C3—C4—C9	-67.47 (17)	C2—C1—C17—C16	-18.7 (2)
C9—C4—C5—O1	177.63 (12)	O1—C1—C17—C16	162.09 (12)
C3—C4—C5—O1	-0.6 (2)	C19—C16—C17—C1	166.24 (13)
C9—C4—C5—C6	-1.5 (2)	C18—C16—C17—C1	-73.32 (17)
C3—C4—C5—C6	-179.70 (13)	C15—C16—C17—C1	46.35 (18)
C1—O1—C5—C4	12.10 (19)	C2—C3—C20—C25	-111.05 (15)
C1—O1—C5—C6	-168.71 (12)	C4—C3—C20—C25	125.02 (15)
C4—C5—C6—C7	2.7 (2)	C2—C3—C20—C21	66.04 (17)
O1—C5—C6—C7	-176.49 (12)	C4—C3—C20—C21	-57.89 (18)
C5—C6—C7—C8	-0.3 (2)	C26—O3—C21—C22	4.4 (2)
C6—C7—C8—C13	176.18 (14)	C26—O3—C21—C20	-175.59 (12)
C6—C7—C8—C9	-3.0 (2)	C25—C20—C21—O3	177.57 (12)
C7—C8—C9—C10	-177.55 (13)	C3—C20—C21—O3	0.4 (2)
C13—C8—C9—C10	3.2 (2)	C25—C20—C21—C22	-2.4 (2)
C7—C8—C9—C4	4.2 (2)	C3—C20—C21—C22	-179.53 (13)
C13—C8—C9—C4	-175.03 (13)	O3—C21—C22—C23	-178.35 (14)
C5—C4—C9—C10	179.85 (12)	C20—C21—C22—C23	1.6 (2)

supplementary materials

C3—C4—C9—C10	-2.0 (2)	C21—C22—C23—C24	0.4 (2)
C5—C4—C9—C8	-2.0 (2)	C22—C23—C24—C25	-1.4 (2)
C3—C4—C9—C8	176.22 (12)	C21—C20—C25—C24	1.3 (2)
C8—C9—C10—C11	-3.2 (2)	C3—C20—C25—C24	178.48 (14)
C4—C9—C10—C11	175.03 (14)	C23—C24—C25—C20	0.6 (2)
C9—C10—C11—C12	0.2 (2)		

Fig. 1

